

TERMS OF USE:

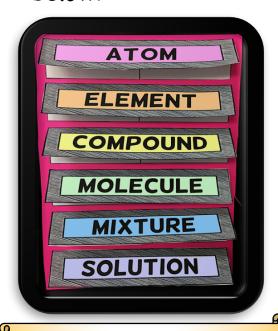
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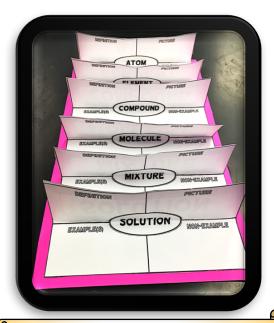
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HOW TO CREATE THE FOLDABLE

- 1. Color the outside of the labels
- 2. Cut out all of the Frayer models and the rectangular labels
- 3. Fold the Frayer models in half, hotdog style, on the middle horizontal line
- Glue the labels (atom, element, compound, etc.) on the top of the corresponding Frayer model rectangle
- 5. Layout all of the pieces on an 8.5×11 paper, making sure the pieces are evenly spaced and that they are in the appropriate order.
- 6. Glue the lower half of each Frayer model to your paper so that the flaps open to show your answers
- 7. The final foldable should look like the images below:

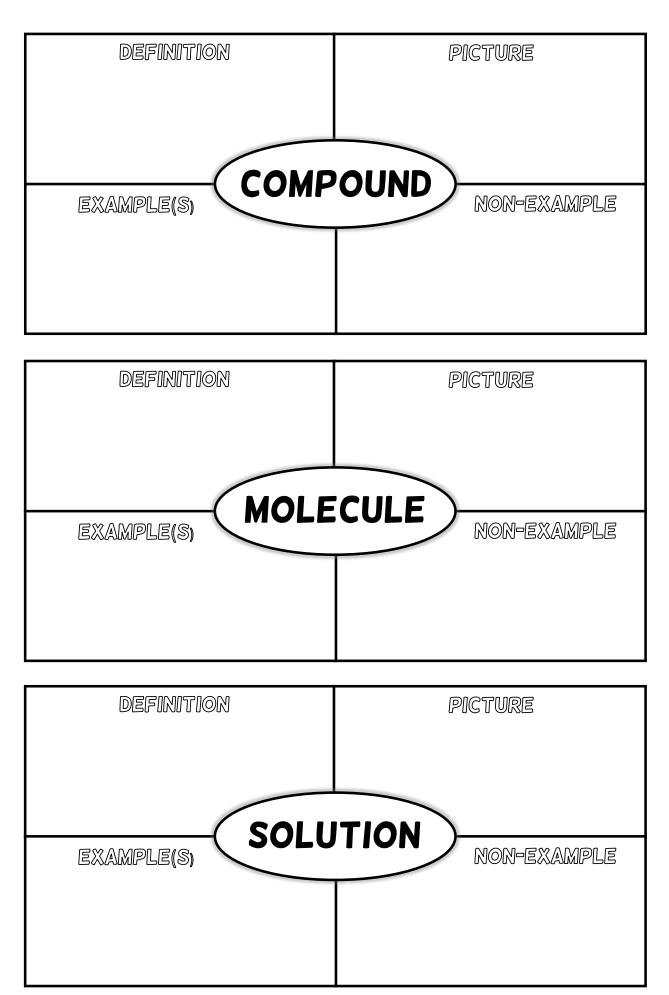


CLOSED FOLDABLE



OPEN FOLDABLE

DEFINITION PICTURE Directions: Cut out each Frayer model and fold hot-dog style on the center, horizontal line. Lay-out all of the folded tabs onto your paper. All six tabs should fit on one page. Place glue onto the back of the boxes (only on the lower half, behind the example/nonexample labels. Glue all six tabs to notebook. Cutout and glue the labels provided onto the outside of each flap. **ATOM** NON-EXAMPLE EXAMPLE(S) DEFINITION PICTURE **ELEMENT** EXAMPLE(S) NON-EXAMPLE DEFINITION PICTURE **MIXTURE** EXAMPLE(S) NON-EXAMPLE



ATOM

ELEMENT

MIXTURE

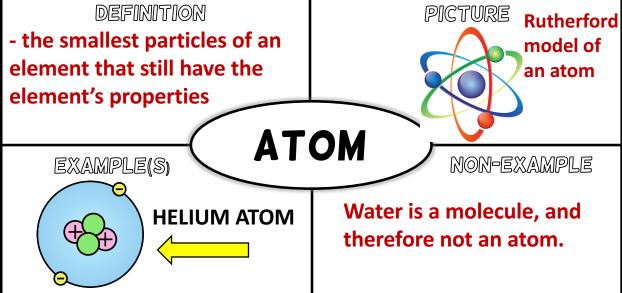
COMPOUND

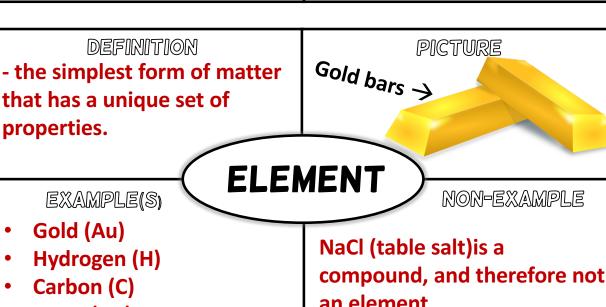
MOLECULE

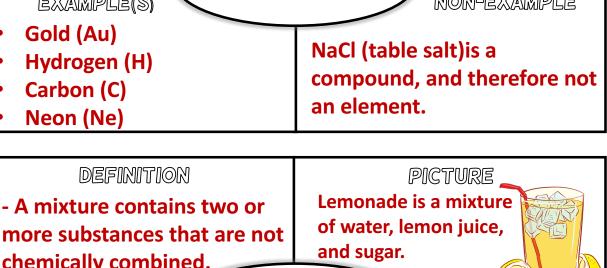
SOLUTION

KEY

Answers will vary, but here is an example of the completed Frayer models. The students should draw their pictures. This key shows clipart so you have an example of what students could possibly draw.







- A mixture contains two or chemically combined.



NON-EXAMPLE

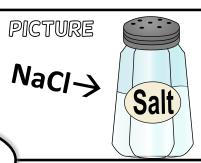
- Homogenous mixture: air, mouthwash, coffee
- **Heterogenous mixture:** cereal, chex mix

EXAMPLE(S)

Silver (Ag) is an element made of only one type of atom, and therefore not a mixture.

DEFINITION

 a substance formed when two or more elements are bonded together.



COMPOUND

non-example

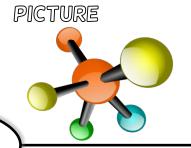
EXAMPLE(S)

- H₂O
- CaCl
- NaHCO₃
- AgBrO₃

Not all molecules are compounds. O3 (ozone) is a molecule but not a compound.

DEFINITION

- the smallest identifiable units of chemical compounds.



NON-EXAMPLE

MOLECULE

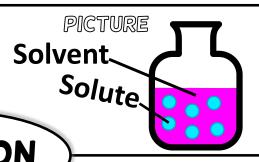
EXAMPLE(S)

- H₂O
- N₂
- CH₄
- O₃

The element Sulfur (S) is not a molecule.

DEFINITION

- a homogenous mixture of one or more solutes dissolved in a solvent.



SOLUTION

EXAMPLE(S)

 Salt (solute) dissolved in water (solvent). A heterogeneous

mixture such as a salad.

MON-EXAMPLE

HELPFUL HINTS

You can use this product in <u>many</u> different ways. Here are a few examples/suggestions of how to use the foldable.

-Students fill out the information for each Frayer Model by conducting their own research using the internet. Or, you can provide students with a science article, or textbook, that supplies the basic information they need. If you do not have a resource for this, I recommend the the science buddies website. Here is a link:

http://www.ck12.org/student/

-The non-examples are always the toughest for students to figure out because they cannot just google the answer. I love the fact that they have to actually APPLY KNOWLEDGE! However, sometimes students will try to take the easy way out and just repeat the other vocab words that are displayed on the foldable. For example, they might write "solution" under the non-example for molecule. I would not give any points for this. Instead, require students to explain why it is a non-example.

-If you have never implemented Frayer Models before, I recommend modeling the appropriate way to fill one out so that they can see how it should be done. You could work the first vocabulary word as a class and then have them complete the rest on their own or even with a partner or as a group activity.

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